

Available online at www.sciencedirect.com



Tetrahedron Letters

Tetrahedron Letters 48 (2007) 203-206

Application of intramolecular Dötz reaction to the synthesis of ansa-compounds: concise synthesis of arnebinol

Masahito Watanabe, Kyosuke Tanaka, Yoko Saikawa* and Masaya Nakata*

Department of Applied Chemistry, Faculty of Science and Technology, Keio University, 3-14-1 Hiyoshi, Kohoku-ku, Yokohama 223-8522, Japan

Received 3 October 2006; revised 9 November 2006; accepted 10 November 2006

Abstract—Fischer carbene complexes having long alkynyloxy chains regioselectively produced oxametacyclophanes via the intramolecular Dötz reaction. The utility of this reaction was demonstrated in the synthesis of arnebinol, an ansa-type terpenoid, from geranyl acetate in six steps.

© 2006 Elsevier Ltd. All rights reserved.

The complex chemical structures and the various biological activities of ansa-compounds (or ansamycins)¹ have attracted great attention in the organic synthetic community. The general synthetic approaches to these ansa-compounds involve elongation of an alkyl chain (an ansa-chain) from an aromatic core followed by macrocyclization at the appropriate position.² As a new synthetic approach to ansa-compounds, we focused on the Dötz reaction.^{3,4} We anticipated that the ansacompounds having oxametacyclophane skeletons, for example, kendomycin (1),⁵ coleophomone A (2),⁶ and arnebinol (3),⁷ would be synthesized featuring the intramolecular Dötz reaction (Fig. 1; bold lines show oxametacyclophane skeletons).⁸ It has been ascertained that the regiochemistry of the intermolecular Dötz reaction using Fischer chromiumcarbene complexes and unsymmetrical alkynes is controlled by unfavorable steric interactions between the large substituent (\mathbb{R}^{L}) of the alkyne and the carbene ligand at the incorporation stage of the alkyne (Scheme 1).⁴ Thus, the major isomer of the obtained *p*-alkoxyphenols has the vicinal \mathbb{R}^{L} and hydroxy substituents. For the intramolecular version of this benzannulation reaction with three types of carbene complexes **4**–**6** bearing the tethered alkynes, the six kinds of products **7–12** are possible; among them, when C* is sterically larger than \mathbb{R}^{3} , compounds **7**, **9**, and **11** would be the major products based on the assumption that the selection rule

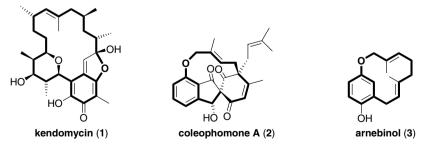


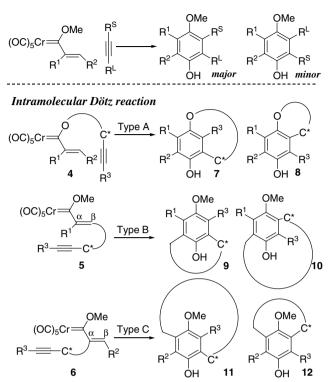
Figure 1. Examples of natural products having oxametacyclophane skeletons.

Keywords: Dötz reaction; Fischer carbene complex; Ansa-compound; Arnebinol.

^{*} Corresponding authors. Tel.: +81 45 566 1562; fax: +81 45 566 1551 (Y.S.); tel.: +81 45 566 1577; fax: +81 45 566 1551 (M.N.); e-mail addresses: saikawa@applc.keio.ac.jp; msynktxa@applc.keio.ac.jp

^{0040-4039/\$ -} see front matter @ 2006 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2006.11.059

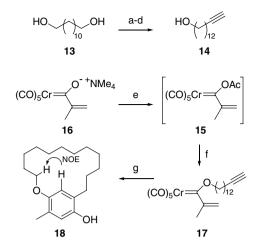
Intermolecular Dötz reaction



Scheme 1. Regiochemistry in inter- and intramolecular Dötz reactions with various chromium–carbene complexes.

in the intermolecular version is applicable. Compound 7 possesses the oxametacyclophane skeleton corresponding to the natural products 1-3. Therefore, we expected that the type A reaction would realize the simultaneous benzannulation and cyclization of the ansa-chain to give the ansa-compound skeleton in one step. The reported examples of the type A, however, show that orthocyclophane 8 was obtained from 4 as the sole product due to the short tether.⁹ In contrast, the β -tethered alkenyl complexes 5 (type B) preferably produced the expected metacyclophane 9.10 A reversal of the regiochemistry of the cyclization was achieved when $R^3 = Ph$ and $C^* = CH_2$ ($R^3 > C^*$), producing paracyclophane 10 as the major isomer.^{10a} The α -tethered alkenyl complexes 6 (type C) showed a behavior similar to the type B cases, giving the expected paracyclophane 11 along with some unexpected rearranged products instead of 12.11 There have been no reports, to the best of our knowledge, about preparing metacyclophane 7 via the type A intramolecular Dötz reaction. It is anticipated that carbene complexes 4 with the long tether length would give oxametacyclophanes 7 corresponding to the desired ansa-compounds. We selected the chromium-carbene complex 17 (vide infra) as the model substrate to verify this hypothesis.

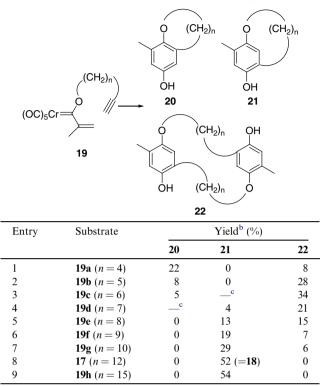
The synthesis of the designed carbene complex 17 started with dodecane-1,12-diol (13) by the four-step transformation into tetradec-13-yn-1-ol (14) (Scheme 2). Alkynyl alcohol 14 was introduced to acetoxyvinyl-idenechromium carbene complex 15,¹² prepared in situ



Scheme 2. Synthesis of the carbene complex 17 and its benzannulation. Reagents and conditions: (a) TrCl (1.0 mol amt), Et₃N (1.5 mol amt), DMAP (0.05 mol amt), CH₂Cl₂, rt, 1 h, 59%; (b) TsCl (1.0 mol amt), Et₃N (1.5 mol amt), DMAP (0.1 mol amt), CH₂Cl₂, rt, 3.5 h, 91%; (c) lithium acetylide ethylenediamine complex (2.0 mol amt), DMSO, rt, 2 h, 89%; (d) camphorsulfonic acid (0.1 mol amt), MeOH– CH₂Cl₂, rt, 2 h, quantitative yield; (e) AcBr (1.1 mol amt), CH₂Cl₂, -78 °C, 1.5 h; (f) 14 (1.0 mol amt), CH₂Cl₂, -78 °C, 3 h, 83%; (g) degassed solvents (0.002 M for 17): 50 °C, 3 h, 29% (hexane): reflux, 3 h, 41% (Et₂O): 50 °C, 3 h, 50% (benzene): 50 °C, 3 h, 52% (toluene). Tr = triphenylmethyl, DMAP = 4-(*N*,*N*-dimethylamino)pyridine, Ts = *p*-toluenesulfonyl.

by treatment of tetramethylammonium salt 16^{13} with acetyl bromide, to give alkynyloxy carbene complex 17. The resulting complex 17 was then subjected to the intramolecular Dötz reaction (50 °C, 3 h) in various solvents. In hexane, Et₂O (reflux), benzene, and toluene, the anticipated oxa-[13]-metacyclophane 18 was obtained as the major product in 29%, 41%, 50%, and 52% yields, respectively. The reaction in THF resulted in a complex mixture. The regiochemistry of 18 was confirmed by the NOE experiment and the absence of the coupling between the aromatic hydrogens in the ¹H NMR studies. From the viewpoint of the vield and reproducibility, we chose toluene as a suitable solvent for further investigations.

A more extensive study of the effect of the tether length was investigated and the results are shown in Table 1. Alkynyloxy carbene complexes 19a-h were prepared by the same method as described above from ammonium salt 16 with the corresponding alkynyl alcohols, which are either commercially available or easily prepared by the isomerization of their internal alkyne isomers.^{14,15} Hexynyloxy carbene complex **19a** produced oxa-[5]-orthocyclophane 20a and dioxa-[5,5]-metacyclophane 22a in 22% and 8% yields, respectively (entry 1). Heptynyloxy carbene complex **19b** favored dimerization rather than orhtocyclization, leading to dioxa-[6,6]metacyclophane 22b in 28% yield (entry 2). With the middle chain (n = 6 and 7) complexes, 19c and 19d, the reaction resulted in a complex mixture, in which the major products were dioxa-[n+1,n+1]-metacyclophanes 22c and 22d (entries 3 and 4). As the alkynyloxy chain became longer (19e-h and 17), the yields of oxa-[n+1]-metacyclophanes **21e**-h (and **18**) increased and Table 1. Intramolecular Dötz reaction of carbene complexes 19a-ha

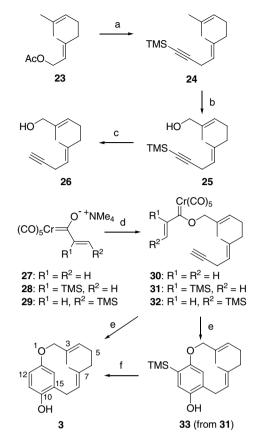


^a All reactions were run at 0.002 M in toluene at 50 °C for 4 h under an argon atmosphere.

^b Isolated yield after silica-gel column chromatography. All reaction mixtures contain some kinds of as-yet unidentified by-products. ^c Not found.

those of the dimers 22e-h decreased (entries 5–9). Apparently, these results suggest that the intramolecular Dötz reaction of 4 (type A) having a sufficient length of the alkynyloxy chain tends to metacyclize to give oxa-[n+1]-metacyclophane 7 in line with the intermolecular preference.

Encouraged by these results, we applied the intramolecular Dötz reaction to the synthesis of the ansa-type monoterpenylbenzenoid arnebinol (3). Arnebinol (3) was isolated from Arnebia euchroma in 19837 and synthesized by Mori's group from geranyl acetate in 1984.¹⁶ Alkynyl alcohol **26** was prepared from geranyl acetate (23) in three steps as follows (Scheme 3). Geranyl acetate (23) was substituted with TMS acetylide in the presence of (DPEphos)PdCl₂ according to Negishi's method¹⁷ to give dienyne **24** in 62% yield. Allylic oxidation of 24 with SeO_2^{18} afforded 25 in 37% yield. Removal of the TMS group was realized using aqueous KF¹⁹ without isomerization of the alkyne to the allene, affording the desired alkynyl alcohol 26 in 98% yield. This was incorporated into the three types of carbene complexes 27-29^{20,21} to afford 30-32, respectively. Among them, both 30 and 32 were not very stable, therefore the isolated yields were low (30% and 26%, respectively); in contrast, 31 was obtained in 82% yield due to its good stability. The benzannulation of 30-32 in toluene at 60 °C showed a different behavior. The unstable 30 produced a fairly good yield of arnebinol (3) (49%); the stable 31 gave 33 (32%) together with a



Scheme 3. Synthesis of arnebinol (3). Reagents and conditions: (a) trimethylsilylacetylene (1.2 mol amt), *n*-BuLi (1.1 mol amt)/hexane, THF, $-78 \,^{\circ}$ C, 0.5 h, then ZnBr₂ (1.1 mol amt), $-78 \,^{\circ}$ C to 0 $^{\circ}$ C, 25 min, then (DPEphos)PdCl₂ (0.005 mol amt), 23 (1.0 mol amt)/DMF, 70 $^{\circ}$ C, 1.5 h, 62%; (b) SeO₂ (0.1 mol amt), TBHP (17 mol amt)/CH₂Cl₂, salicylic acid (0.1 mol amt), CH₂Cl₂, rt, 6 h, 37%; (c) KF (5.5 mol amt), H₂O (4.5 mol amt), DMF, rt, 16 h, 98%; (d) 27, AcBr (1.1 mol amt), CH₂Cl₂, $-78 \,^{\circ}$ C, 0.5 h, then 26, $-78 \,^{\circ}$ C o $^{\circ}$ C to rt, 82% (for 31): 29, AcBr (1.2 mol amt), CH₂Cl₂, $-78 \,^{\circ}$ C, 1.5 h, then 26, $-78 \,^{\circ}$ C, 1.5 h, then 26, $-78 \,^{\circ}$ C, 6 h, 49% (from 30): 60 $^{\circ}$ C, 3 h, 32% (from 31 to 33): 60 $^{\circ}$ C, 3 h, 21% (from 32); (f) TFA (9.3 mol amt), CH₂Cl₂, $0 \,^{\circ}$ C, 0.5 h, 83%. DPEphos = bis(2-diphenylphosphinophenyl)ether, TBHP = *tert*-butyl-hydroperoxide.

significant amount of the cyclic dimer (13%). The resulting **33** was desilylated with TFA to give arnebinol (**3**) in 83% yield. The unstable **32** directly gave **3** accompanied by the silyl C \rightarrow O migration and desilylation,²⁰ allbeit in a low yield (21%). No regioisomer (oxa-[9]-orthocyclophane) was observed in all the benzannulation products. From the viewpoint of sample-handling easiness and overall yield, the route of **28** \rightarrow **31** \rightarrow **33** \rightarrow **3** was adopted. The analytical and spectroscopic data of the synthetic sample of arnebinol (**3**) matched those of the natural product⁷ and Mori's results.^{16,22}

In summary, we described here the fundamental investigation of the type A intramolecular Dötz reaction and the application of this reaction to the regioselective synthesis of arnebinol (six steps). Syntheses of more functionalized ansa-compounds using this reaction are now in progress in our laboratories.

References and notes

- For recent reviews on ansamycins, see: (a) Funayama, S.; Cordell, G. A. In *Studies in Natural Products Chemistry*; Atta-ur-Rahman, Ed.; Elsevier Science: Amsterdam, 2000; Vol. 23, pp 51–106; (b) Floss, H. G.; Yu, T.-W. *Chem. Rev.* 2005, 105, 621–632; (c) Bryskier, A. In *Antimicrobial Agents*; Bryskier, A., Ed.; ASM Press: Washington, 2005; pp 906–924.
- (a) Isobe, M. In *Recent Progress in the Chemical Synthesis of Antibiotics*; Lukacs, G., Ohno, M., Eds.; Springer-Verlag: Berlin, 1990; pp 103–134; (b) Snader, K. M. In *Anticancer Agents from Natural Products*; Cragg, G. M., Kingston, D. G. I., Newman, D. J., Eds.; CRC Press: Boca Raton, 2005; pp 339–356.
- 3. Dötz, K. H. Angew. Chem., Int. Ed. Engl. 1975, 14, 644-645.
- For reviews on Fischer carbene complexes, see: (a) Dötz, K. H.; Tomuschat, P. Chem. Soc. Rev. 1999, 28, 187–198; (b) de Meijere, A.; Schirmer, H.; Duetsch, M. Angew. Chem., Int. Ed. 2000, 39, 3964–4002; (c) Semmelhack, M. F. In Organometallics in Synthesis: A manual; Schlosser, M., Ed.; John Wiley & Sons: Chichester, 2002; pp 1003– 1121; (d) Dötz, K. H.; Minatti, A. In Transition Metals for Organic Synthesis; Beller, M., Bolm, C., Eds.; Wiley-VCH Verlag GmbH & Co. KGaA: Weinheim, 2004; pp 397– 425.
- (a) Funahashi, Y.; Kawamura, N.; Ishimaru, T. Jpn. Kokai Tokkyo Koho 1996, 08231551. Chem. Abstr. 1996, 125, 326518; (b) Funahashi, Y.; Kawamura, N.; Ishimaru, T. Jpn. Kokai Tokkyo Koho 1996, 08231552. Chem. Abstr. 1997, 126, 6553; (c) Su, M. H.; Hosken, M. I.; Hotovec, B. J.; Johnston, T. L. U.S. Patent 5728727, 1998; Chem. Abstr. 1998, 128, 239489; (d) Bode, H. B.; Zeeck, A. J. Chem. Soc., Perkin Trans. 1 2000, 323–328.
- (a) Kamigauchi, T.; Nakajima, M.; Tani, H. Jpn. Kokai Tokkyo Koho 1998, 10101666. Chem. Abstr. 1998, 129, 589; (b) Wilson, K. E.; Tsou, N. N.; Guan, Z.; Ruby, C. L.; Pelaez, F.; Gorrochategui, J.; Vicente, F.; Onishi, H. R. Tetrahedron Lett. 2000, 41, 8705–8709.
- Xin-Sheng, Y.; Ebizuka, Y.; Noguchi, H.; Kiuchi, F.; Iitaka, Y.; Sankawa, U.; Seto, H. *Tetrahedron Lett.* 1983, 24, 2407–2410.
- White and Smits reported the partial synthesis of kendomycin using the intermolecular Dötz reaction, see: White, J. D.; Smits, H. Org. Lett. 2005, 7, 235–238.
- (a) Semmelhack, M. F.; Bozell, J. J. Tetrahedron Lett. 1982, 23, 2931–2934; (b) Semmelhack, M. F.; Bozell, J. J.; Sato, T.; Wulff, W.; Spiess, E.; Zask, A. J. Am. Chem. Soc. 1982, 104, 5850–5852; (c) Semmelhack, M. F.; Bozell, J. J.; Keller, L.; Sato, T.; Spiess, E. J.; Wulff, W.; Zask, A. Tetrahedron 1985, 41, 5803–5812; (d) Peterson, G. A.; Kunng, F.-A.; McCallum, J. S.; Wulff, W. D. Tetrahedron Lett. 1987, 28, 1381–1384; (e) Wulff, W. D.; McCallum, J. S.; Kunng, F.-A. J. Am. Chem. Soc. 1988, 110, 7419–7434; (f) Balzer, B. L.; Cazanoue, M.; Finn, M. G. J. Am. Chem. Soc. 1992, 114, 8735–8736; (g) Gross, M. F.; Finn, M. G. J. Am. Chem. Soc. 1994, 116, 10921–10933.
- (a) Wang, H.; Wulff, W. D. J. Am. Chem. Soc. 1998, 120, 10573–10574; (b) Dötz, K. H.; Gerhardt, A. J. Organomet. Chem. 1999, 578, 223–228; (c) Dötz, K. H.; Mittenzwey, S. Eur. J. Org. Chem. 2002, 39–47.

- (a) Wang, H.; Wulff, W. D.; Rheingold, A. L. J. Am. Chem. Soc. 2000, 122, 9862–9863; (b) Wang, H.; Huang, J.; Wulff, W. D.; Rheingold, A. L. J. Am. Chem. Soc. 2003, 125, 8980–8981.
- Fischer, E. O.; Selmayr, T.; Kreißl, F. R. Chem. Ber. 1977, 110, 2947–2955.
- (a) Fischer, E. O.; Maasböl, A. Chem. Ber. 1967, 100, 2445–2456; (b) Chamberlin, S.; Wulff, W. D.; Bax, B. Tetrahedron 1993, 49, 5531–5547.
- (a) Brown, C. A.; Yamashita, A. J. Am. Chem. Soc. 1975, 97, 891–892; (b) Lindhoudt, J. C.; van Mourik, G. L.; Pabon, H. J. J. Tetrahedron Lett. 1976, 17, 2565–2568; (c) Abrams, S. R.; Nucciarone, D. D.; Steck, W. F. Can. J. Chem. 1983, 61, 1073–1076.
- 15. Dodec-11-yn-1-ol (for 19g) was synthesized by the same method as that in the case of 14. Heptadec-16-yn-1-ol (for 19h) was prepared by coupling of propargyl alcohol and 1-bromotetradecane followed by isomerization.¹⁴
- (a) Mori, K.; Sakakibara, M.; Waku, M. Tetrahedron Lett.
 1984, 25, 1085–1086; (b) Mori, K.; Waku, M.; Sakakibara, M. Tetrahedron 1985, 41, 2825–2830.
- 17. Qian, M.; Negishi, E. Synlett 2005, 1789-1793.
- Umbreit, M. A.; Sharpless, K. B. J. Am. Chem. Soc. 1977, 99, 5526–5528.
- (a) Corey, E. J.; Fleet, G. W. J.; Kato, M. *Tetrahedron Lett.* **1973**, *14*, 3963–3966; (b) Balme, G.; Bouyssi, D. *Tetrahedron* **1994**, *50*, 403–414; (c) Aïssa, C.; Dhimane, A.-L.; Malacria, M. *Synlett* **2000**, 1585–1588.
- Chamberlin, S.; Wulff, W. D. J. Org. Chem. 1994, 59, 3047–3054, and references cited therein.
- Carbene complexes 27–29 were prepared by the standard Fischer method^{13a} from Cr(CO)₆ and the corresponding vinyl lithium (tetravinyltin, *n*-BuLi (for 27): (α-bromovinyl)trimethylsilane, *t*-BuLi (for 28): trimethylsilylacetylene, (*n*-Bu)₃SnH, AIBN, 80 °C, 2 d, then *n*-BuLi/hexane, -78 °C to rt (for 29)). Preparation of 29, see: (a) Cunico, R. F.; Clayton, F. J. J. Org. Chem. 1976, 41, 1480–1482; (b) Challener, C. A.; Wulff, W. D.; Anderson, B. A.; Chamberlin, S.; Faron, K. L.; Kim, O. K.; Murray, C. K.; Xu, Y.-C.; Yang, D. C.; Darling, S. D. J. Am. Chem. Soc. 1993, 115, 1359–1376.
- 22. The reported assignments⁷ of ¹³C NMR spectrum were partially revised. Arnebinol (3): a colorless prism; mp 163.5–165.8 °C (benzene) (lit.⁷ 163.5–164 °C (benzene), lit.^{16a} 159–160 °C (benzene), lit.^{16b} 161.5–163.0 °C (benzene) ene)); UV (EtOH) λ_{max} nm (log ε): 214.5 (4.34), 292 (3.53) [lit.⁷ 213.2 (4.42), 294.2 (3.83), lit.^{16b} 203 (4.36)]; IR (KBr) v_{max} cm⁻¹: 3423, 2941, 2905, 2840, 1507, 1458, 1440, 1420, 1342, 1253, 1192, 1145, 988, 850, 803; ¹H NMR (CDCl₃, CHCl₃ = 7.26): δ 1.25 (3H, s, 7-Me), 1.51 (3H, s, 3-Me), 2.14 (1H, br s, H-5), 2.34 (2H, t, J = 7.0 Hz, H-6), 2.49 (1H, br s, H-5), 3.07 (1H, br s, H-9), 3.30 (1H, br s, H-9), 4.32 (1H, s, OH), 4.54 (2H, br s, H-2), 5.53 (1H, br t, J = 7.0 Hz, H-4), 5.68 (1H, br t, J = 7.0 Hz, H-8), 6.56 (1H, dd, J = 8.5, 2.6 Hz, H-12), 6.60 (1H, d, J = 8.5 Hz, H-11), 7.45 (1H, d, J = 2.6 Hz, H-14); ¹³C NMR (CDCl₃, $CDCl_3 = 77.00$): δ 12.64 (3-Me), 14.76 (7-Me), 25.28 (C5), 25.77 (C9), 39.24 (C6), 76.58 (C2), 115.03 (C11 or C12), 115.29 (C12 or C11), 118.13 (C14), 124.08 (C8), 127.08 (C15), 132.56 (C3), 133.64 (C4), 140.30 (C7), 146.84 (C10), 151.75 (C13); HR-EI-MS m/z calcd for $C_{16}H_{20}O_2$ ([M]⁺) 244.1463, found 244.1460.